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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.	
10/586,470	07/18/2006	Tim Jungkamp	12810-00322-US1	4266	
	7590 02/03/201 OVE LODGE & HUT		EXAM	IINER	
1875 EYE STREET, N.W. KOSACK, JOSEPH R			JOSEPH R		
SUITE 1100 WASHINGTO	N, DC 20006		ART UNIT PAPER NUMBER		
			1626		
			MAIL DATE	DELIVERY MODE	
			02/03/2011	PAPER	

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

	Application No.	Applicant(s)	
	10/586,470	JUNGKAMP ET AL.	
Office Action Summary	Examiner	Art Unit	
	Joseph R. Kosack	1626	
The MAILING DATE of this communication ap Period for Reply	pears on the cover sheet wit	h the correspondence address	s
A SHORTENED STATUTORY PERIOD FOR REPL WHICHEVER IS LONGER, FROM THE MAILING D. - Extensions of time may be available under the provisions of 37 CFR 1. after SIX (6) MONTHS from the mailing date of this communication. - If NO period for reply is specified above, the maximum statutory period Failure to reply within the set or extended period for reply will, by statut Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	DATE OF THIS COMMUNIC 136(a). In no event, however, may a re will apply and will expire SIX (6) MONT e, cause the application to become ABA	CATION. ply be timely filed "HS from the mailing date of this commun ANDONED (35 U.S.C. § 133).	
Status			
Responsive to communication(s) filed on 19 № This action is FINAL . 2b) This 3) Since this application is in condition for allowed closed in accordance with the practice under	s action is non-final. ance except for formal matte	•	rits is
Disposition of Claims			
 4) Claim(s) 1-20 is/are pending in the application 4a) Of the above claim(s) 11-16 is/are withdra 5) Claim(s) is/are allowed. 6) Claim(s) 1-10 and 17-20 is/are rejected. 7) Claim(s) is/are objected to. 8) Claim(s) are subject to restriction and/orange. 	wn from consideration.		
Application Papers			
9) The specification is objected to by the Examina 10) The drawing(s) filed on is/are: a) accomposed as a specific at any objection to the Replacement drawing sheet(s) including the correct and the oath or declaration is objected to by the Examination.	cepted or b) objected to be drawing(s) be held in abeyand ction is required if the drawing(s	ce. See 37 CFR 1.85(a). s) is objected to. See 37 CFR 1.	
Priority under 35 U.S.C. § 119			
12) Acknowledgment is made of a claim for foreign a) All b) Some * c) None of: 1. Certified copies of the priority documen 2. Certified copies of the priority documen 3. Copies of the certified copies of the priority documen application from the International Burea * See the attached detailed Office action for a list	its have been received. Its have been received in Apprity documents have been Bu (PCT Rule 17.2(a)).	oplication No received in this National Stag	e
Attachment(s)			
1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date	Paper No(s	ummary (PTO-413) //Mail Date formal Patent Application _	

DETAILED ACTION

Claims 1-20 are pending in the instant application.

Claim Rejections - 35 USC § 103

Claims 1-10 and 17-20 were previously rejected under 35 U.S.C. 103(a) as being unpatentable over Drinkard et al. (USPN 3,356,748) in view of Fischer et al. (USPN 6,242,633) and Jungkamp et al. (WO 02/26698).

The Applicant's arguments have been considered, but have not been found to be persuasive for the following reasons.

Firstly, while Jungkamp et al. do not teach a distillation example with the exact pairs of pentenenitrile isomers as in claim 1 and the dependent claims, Jungkamp et al. appreciates that the claimed isomers would be in a reaction mixture generated from the direct hydrocyanation of 1,3-butadiene with hydrogen cyanide. See page 1, lines 40-47.

Jungkamp et al. teaches that by azeotropic distillation, groups of pentenenitrile isomers can be separated because their relative volatility ratio (alpha) will be higher than without the addition of water. See page 1, lines 6-13. The implication is that the addition of water, or another diluent to form an azeotrope, is not specifically necessary if the relative volatility ratio (alpha) is higher than 1.3. The relative volatility between Z-2M2BN and 2M3BN is 1.12. Therefore, from looking at the teachings from Jungkamp et al. shown above, Jungkamp et al. would suggest that azeotropic distillation would be effective at separating this pair of isomers. Additionally, the claims do not exclude the addition of water to the mixtures as the mixtures are described in open language by

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virtue of the term "comprising." Therefore, additional components, such as water, can be present.

Jungkamp et al. additionally clearly suggest that the distillation can be accomplished under reduced pressure. While the most preferred way of distillation described in Jungkamp et al. is at atmospheric pressure, Jungkamp et al. teaches that the distillation can be carried out advantageously at a pressure ranging from 1 to 200 kPa. See page 5, lines 24-26. As 1 bar is equal to 100 kPa, the range taught by Jungkamp et al. clearly contains the instantly claimed range of pressures.

Even though the Applicant has shown in the instant specification that distillation efficiency is increased at lower pressures, the Examiner believes that the suggestion by Jungkamp et al. that the distillation may be done at lower pressures and the overall prima facie case for obviousness is stronger than the evidence of increased efficiency since any vacuum distillation apparatus may be used in the instantly claimed process.

Therefore, the Examiner has considered the Applicant's arguments fully, but has not found them to be persuasive. The rejection is maintainted

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

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The factual inquiries set forth in *Graham* **v.** *John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:

- 1. Determining the scope and contents of the prior art.
- 2. Ascertaining the differences between the prior art and the claims at issue.
- 3. Resolving the level of ordinary skill in the pertinent art.
- 4. Considering objective evidence present in the application indicating obviousness or nonobviousness.

This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).

Claims 1-10 and 17-20 rejected under 35 U.S.C. 103(a) as being unpatentable over Drinkard et al. (USPN 3,356,748) in view of Fischer et al. (USPN 6,242,633) and Jungkamp et al. (WO 02/26698).

The claims are drawn to a process for preparing 3-pentenenitrile by isomerizing 2-methyl-3-butenenitrile over a catalyst and distilling the products away from each other. A dependent claim details that the 2-methyl-3-butenenitrile is generated by hydrocyanation of 1,3-butadiene and separating the reaction products by distillation.

Drinkard et al. teach the isomerization of 2-methyl-3-butenenitrile over a tetrakis(triethyl phosphite) nickel(0) catalyst in order to generate 3-pentenenitrile. See column 4, Example 1.

Drinkard et al. do not teach where the product nitriles are separated from each other by distillation and where the reactant stream comes from the hydrocyanation of 1,3-butadiene.

Fischer et al. teach the hydrocyanation reaction of 1,3-butadiene with a nickel phosphite catalyst to form pentenenitriles which include 2-methyl-3-butenenitrile. See Example 15, column 21.

Jungkamp et al. teach the azeotropic distillation of various pentenenitrile isomers with the pressures and temperatures required by the claims. See page 2, line 39 through page 3, line 6 and Table 1, page 7. Jungkamp et al. do not teach the exact pairs of isomers that are listed in claim 1 nor does Jungkamp et al. teach the exact reactions that the mixtures come from.

It would be obvious to one of ordinary skill to take the method proven by Jungkamp et al. and apply it to other mixtures of pentenenitrile isomers as distillation techniques such as simple distillation, fractional distillation, vacuum distillation, and azeotropic distillation are well known in the art and are readily applied by the person of ordinary skill in purifying isomeric liquids from one another. As to the reaction that the mixtures come from, one of skill in the art would be able to complete the distillation irrespective for which reaction the mixture of pentenenitriles originated from.

Therefore the claims are *prima facie* obvious over the prior art.

Conclusion

Claims 1-10 and 17-20 are rejected.

THIS ACTION IS MADE FINAL. Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Joseph R. Kosack whose telephone number is (571)272-5575. The examiner can normally be reached on M-Th 6:30-5:00.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Joseph McKane can be reached on (571)-272-0699. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/Joseph R Kosack/ Primary Examiner, Art Unit 1626